

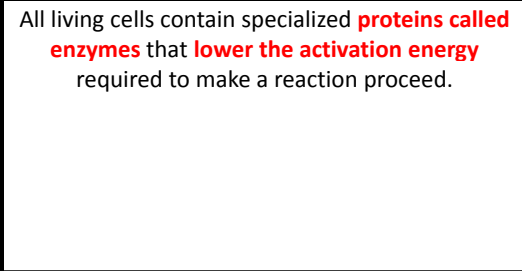
Enzymes and Energy

To release chemical energy to perform work cells must have a way to break and form chemical bonds.



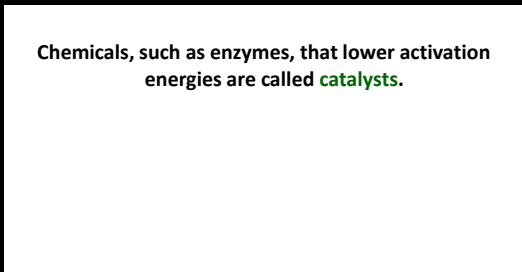
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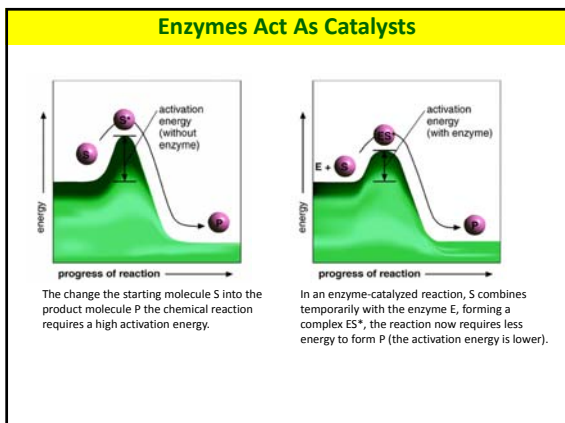
All living cells contain specialized **proteins called enzymes** that **lower the activation energy** required to make a reaction proceed.

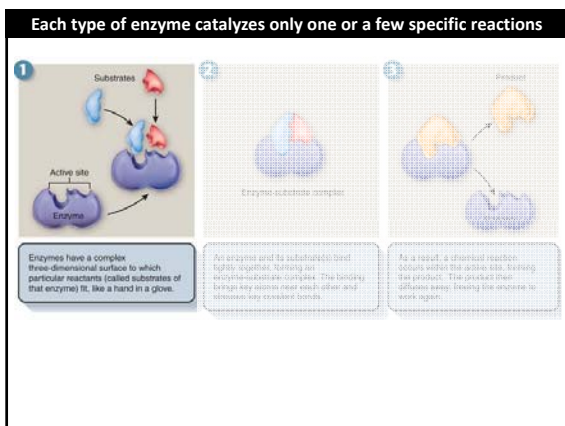


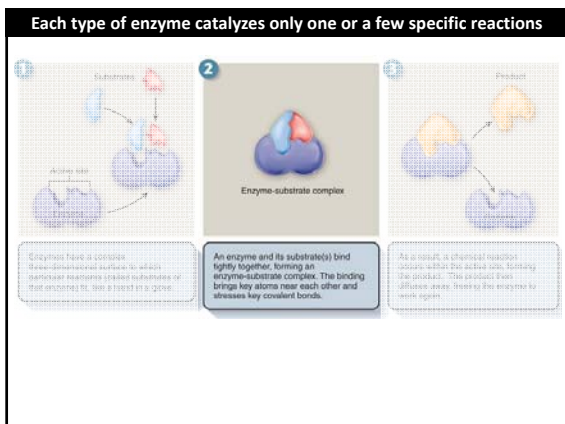
Enzymes and Energy

Chemicals, such as enzymes, that lower activation energies are called **catalysts**.









Each type of enzyme catalyzes only one or a few specific reactions

1 Substrates
Active site
Enzyme

Enzymes have a complex three-dimensional surface to which particular reactants (called substrates of that enzyme) fit, like a hand in a glove.

2 Enzyme-substrate complex

An enzyme and its substrate(s) bind tightly together, forming an enzyme-substrate complex. The binding brings key atoms near each other and stresses key covalent bonds.

3 Product
Enzyme

As a result, a chemical reaction occurs within the active site, forming the product. The product then diffuses away, freeing the enzyme to work again.

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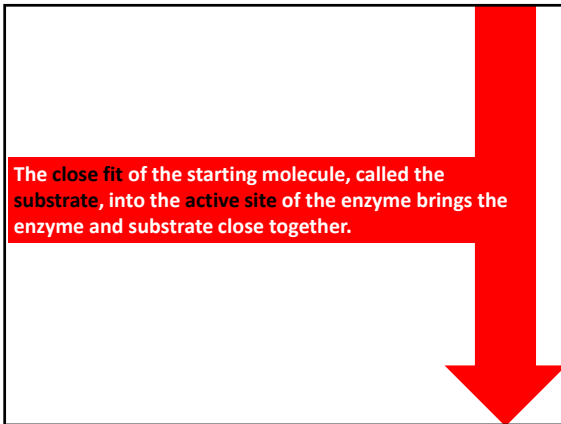
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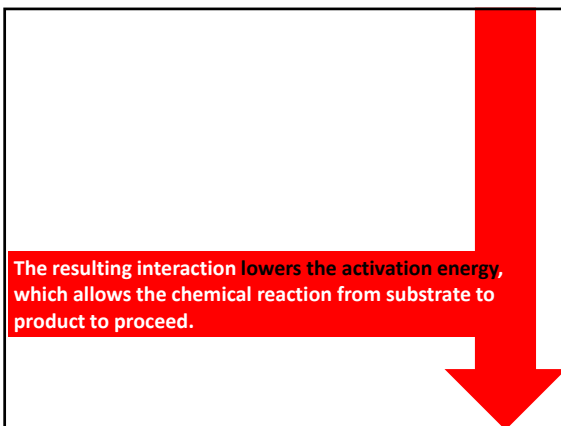
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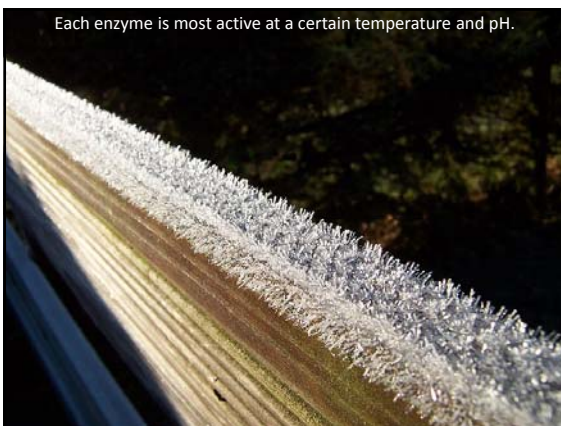
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Enzyme

As a result, a chemical reaction occurs within the active site, forming the product. The product then diffuses away, freeing the enzyme to work again.

The specific reaction catalyzed by an enzyme depends on a small area of its structure called the active site.







The **pH** of a solution

- pH is used to indicate the acidity of a solution
- pH has values that usually range from 0 to 14
- pH is acidic when the values are less than 7
- pH is neutral with a pH of 7
- pH is basic when the values are greater than 7



Learning Check

Identify each solution as A) acidic, B) basic, or N) neutral

- ___ 1) HCl with a pH = 1.5
- ___ 2) pancreatic fluid pH = 7.9
- ___ 3) Sprite pH = 3.0
- ___ 4) pH = 7.0

Identify each solution as A) acidic, B) basic, or N) neutral

- A) 1) HCl with a pH = 1.5
- B) 2) Pancreatic fluid pH = 7.9
- A) 3) Sprite soft drink pH = 3.0
- N) 4) pH = 7.0

The pH of solutions can be determined using

- a pH meter
- pH paper
- indicators that have specific colors at different pH values